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Quantifying the Long-Term Stability of Food Formulations

Foodstuffs are often mixtures of different ingredients and are therefore liable to separate. It is important to quantify these separation processes to optimize food product formulations. The MultiScan MS 20 dispersion stability analysis system, developed by DataPhysics Instruments, can reliably and efficiently analyze food dispersions regarding their destabilization and aging. The following use case evaluates five milk substitutes and compares the occurring destabilization mechanisms.

by Dr. Qiongjie Liu

Many food products, such as milk products and mayonnaise, are complex multicomponent mixtures. In their production, a homogeneous composition with good flavor, texture, and nutritional value is desired. Separation processes in these mixtures can strongly influence their taste and mouthfeel. Assuring long-term homogeneity and thus a long shelf life for such products requires thorough optimization during the formulation of the mixtures.

The type and concentration of the formulation's ingredients, its processing temperature, pH, and viscosity are crucial parameters determining its separation behavior. Therefore, understanding the stability of food dispersions is important for research and development departments in the food industry. However, observing separation processes with the naked eye is

highly subjective and time-consuming, hindering efficient formulation optimization.

MultiScan MS 20: An Ideal Companion to Characterize Dispersions

To solve the above issues, the German measuring equipment manufacturer DataPhysics Instruments has developed the MultiScan MS 20 measuring device for automatic optical stability and ageing analysis of liquid dispersions (Figure 1). The device was particularly developed to characterize suspensions was well as emulsions and allow comprehensive time- and temperature-resolved analysis of destabilization mechanisms.

The MultiScan MS 20 comprises a base unit to which up to six sample chambers can be added. The base unit of the MultiScan MS 20 features an integrated touch screen, which

displays status information and can be used to control important functions. Additionally, samples can be registered quickly and conveniently with a built-in scanner as well as an optional handheld barcode scanner. The six sample chambers of the MultiScan MS 20, called ScanTowers, can be individually controlled and operated at different temperatures. Electric heating, with liquid counter-cooling, enables measurements between -10 °C and 80°C

Visualising and Quantifying Destabilization Processes

With a MultiScan MS 20, destabilization mechanisms in dispersions can be visualized and quantified. The device achieves this by tracking changes in light intensity. It measures the sample repeatedly for a predefined period and in a position-resolved manner. The patterns arising in the light intensity can give a trained scientist information on how the dispersion changes over time and what destabilization mechanisms might cause these changes.

The experimental setup is as follows: A sample is placed in a glass vial, which is then placed in a measurement chamber. Two light sources and a light detector simultaneously move up and down along the sample (Figure 3). During the measurement, the whole sample height is scanned, detecting global and local changes in the sample. The software analyzes the light intensities transmitted through and backscattered by the sample. The transmission (light penetrating through the sample) and backscattering (light reflected by the sample) intensities depend directly on the number, size, and type of the dispersed particles. Hence, the light

Figure 1: DataPhysics Instruments' MultiScan MS 20 dispersion stability analysis system can be fitted with up to six sample chambers.





Figure 2: The five milk samples, pictured here at the end of the experiment

intensity changes when the dispersion destabilizes.

Particles can sediment to the bottom of the sample container, so more particles scatter back light in the bottom layer. Particles can also cream, meaning lighter particles collect at the top of the liquid, meaning more light is scattered back in the top layer of the dispersion. In both cases, the measured profile of the dispersion decreasing shows transmission intensities, while the backscattering intensities increase in the respective area of the dispersion. Moreover, particles can cluster, minimising their interfacial area with the surrounding liquid phase. Such changes can be seen as global changes in the transmission and backscattering intensities.

Use Case: Vegan Milk Substitutes

The following use case compares the destabilization mechanisms of five milk substitutes using the MultiScan MS 20. Milk substitutes are generally produced by mechanically breaking down plant materials, such as nuts, legumes, or seeds. The plant material is then mixed with oil, water, and colloidal matter to form a multiphase dispersion. This study characterizes five milk substitutes based on peas, oats, coconuts, almonds, and soy (Figure 2).

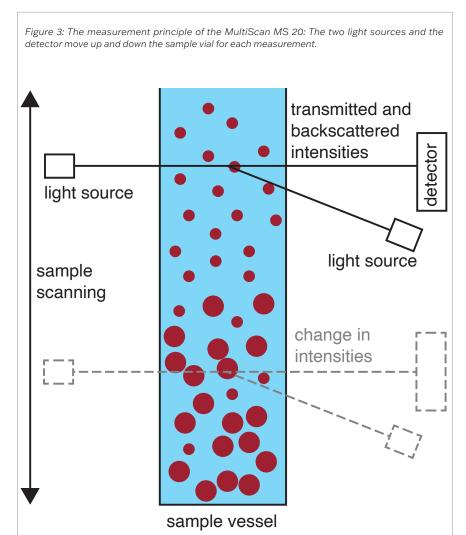
A sample of each substitute with a volume of 20 ml was homogenized using a shaker and then poured into a transparent glass vial to be measured at a constant temperature of 25 °C. The mixtures were measured every 6 minutes for a total experiment time of 1 day and 3 hours. The measured

zone was between 0 mm (bottom of the glass) and 57 mm (top of the glass).

As the samples were opaque, the light cannot penetrate through the sample. Thus, the transmission intensities could not be used for evaluation. Therefore, the backscattering intensities were analyzed to study the stability of all five milk substitutes. The graphs in Figure 4 show the backscattering intensities

against their height positions for all five samples. The color-coding of the curves indicates the time they were recorded, from red (start of the experiment, t = 0 s) to purple (end of the experiment, t = 1 day 3 hours). Each curve represents one measurement.

The backscattering diagrams show time- and position-dependent changes in the light intensities in all five samples. For example, the backscattering intensity of the oat-based substitute decreases at the bottom of the vial while increasing at the top over time. Thus, the particles migrate from the bottom layer to the top layer in a mechanism described as creaming. Regarding the coconut- and almondbased samples, the backscattering intensities at the top also increase; however, so do the backscattering values at the bottom of the sample vial. In this case, creaming at the top is accompanied by a sedimentation



process at the bottom of the vial. Additionally, the middle layer clears as particles migrate to either the top or the bottom. The different assumed destabilization mechanisms are summarised in Table 1.

The software accompanying the MultiScan MS 20 enables a more detailed observation of specific areas to quantify the changes occurring. Figure 5 shows how the cream thickness in the top layer of the sample changes over time. The creaming processes in the oat-, coconut-, and almond-based drinks started instantly and were completed after approximately 2.5 hours. Conversely, the cream layers in the soy- and pea-based drinks formed much later. The cream layer in the oat-based drink was the

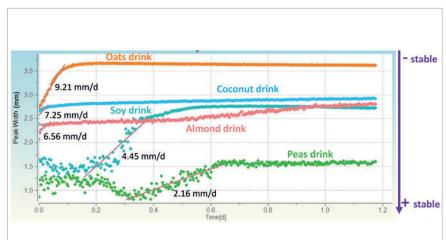
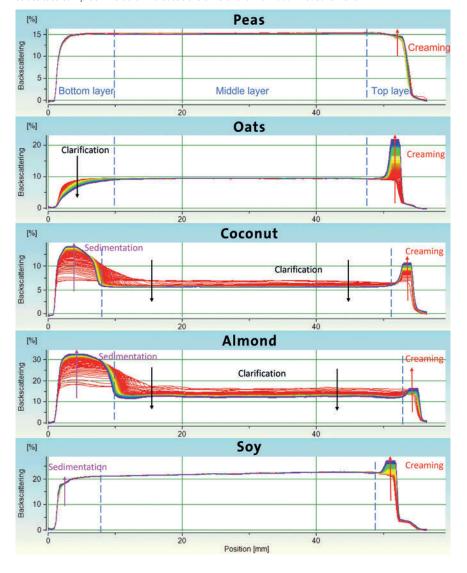


Figure 5: The graph compares the thicknesses of the creaming layers at the top of the samples for all samples versus time.

thickest, with a height of around 3.6 mm, whereas that in the pea-based drink was the thinnest at around 1.5 mm.

Figure 4: The graphs show the backscattering intensity versus the height position for the five milk substitute samples. The color indicates the time of the individual measurement.



Additionally, the software can reveal the sedimentation kinetics of the samples. The so-called 'peak-area change rate' of the almond-based drink was highest with a value of 418.8 mm%/d, whereas the change rate of the pea-based drink was lowest with a value of only 1.72 mm%/d. These results indicate that particles in the pea-based drink sediment much more slowly than those in the almond-based drink.

The software can also provide a global stability analysis using the stabilityindex (SI) function. This function summarises and quantifies the effects of various destabilization mechanisms over the entire sample height. With these SI values, the stability of different products can be compared directly (Figure 6). Consistent with the changes in backscattering intensities, the peabased drink was found to be the most stable formulation, while the stability of the almond drink was the lowest. These results underline the excellent applicability of the MultiScan MS 20 to analyze and quantify stability issues of different formulations with high reliability.

Conclusion

Destabilization processes occur in most multicomponent mixtures, including food products. Such processes strongly influence the product's final appearance, taste, and feel. Therefore, it is essential for research and development departments in the food industry to quantify and analyze the destabilization processes of multicomponent mixtures. Such

Sample	Bottom Layer	Middle Layer	Top Layer
Pea	-	-	Creaming
Oat	Clarification	-	Creaming
Coconut	Sedimentation	Clarification	Creaming
Almond	Sedimentation	Clarification	Creaming
Soy	Sedimentation	-	Creaming

Table 1: Possible destabilization mechanisms for the five milk substitutes, broken down into top, middle and bottom layer

evaluations enable efficient optimization of the product formulations. Using the MS 20 dispersion stability analysis system and its corresponding software, a quantitative and exact means to study the stability of multicomponent mixtures, such as plant-based milk substitutes, was shown. With this device, changes in light intensity can be detected readily and reliably, enabling producers to anticipate and quantify stability issues and thus allow time- and cost-optimized product development.

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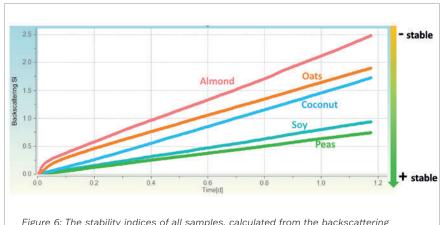


Figure 6: The stability indices of all samples, calculated from the backscattering intensities versus time





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